

Quiz 8

1. (5 points). Compute the derivative of $H(z) = \ln \sqrt{\frac{a^2 - z^2}{a^2 + z^2}}$.

By using rules for logarithms, we may rewrite H as

$$\begin{aligned} H(z) &= \ln \sqrt{\frac{a^2 - z^2}{a^2 + z^2}} \\ &= \frac{1}{2} \ln \left(\frac{a^2 - z^2}{a^2 + z^2} \right) \\ &= \frac{1}{2} (\ln(a^2 - z^2) - \ln(a^2 + z^2)). \end{aligned}$$

With this, it becomes easy to compute the derivative:

$$\begin{aligned} H'(z) &= \frac{1}{2} \left(\frac{-2z}{a^2 - z^2} - \frac{2z}{a^2 + z^2} \right) \\ &= \frac{-z(a^2 + z^2) - z(a^2 - z^2)}{a^4 - z^4} \\ &= \frac{-2a^2z}{a^4 - z^4}. \end{aligned}$$

Note that H is a function of z and z only (which is why it is written $H(z)$). That is, a is a constant.

It is also possible to compute the derivative directly using the chain rule, however it takes more work to simplify the result.

2. (5 points). Compute the derivative of $f(x) = (x^2 - 1)^{\sin x}$.

This function has an exponential with a base and exponent which are both varying with x , so our normal methods do not apply. There are two ways to compute this derivative (both substantially the same). The first is by logarithmic differentiation. Write $y = (x^2 - 1)^{\sin x}$ and take the logarithm of both sides to obtain $\ln y = \sin(x) \ln(x^2 - 1)$. Then, we implicitly differentiate each side,

$$\frac{1}{y} \frac{dy}{dx} = \sin(x) \frac{2x}{x^2 - 1} + \cos(x) \ln(x^2 - 1),$$

and multiply by y to solve for $\frac{dy}{dx}$:

$$\frac{dy}{dx} = \left(\frac{2x \sin(x)}{x^2 - 1} + \cos(x) \ln(x^2 - 1) \right) \sin(x) \ln(x^2 - 1).$$

Alternatively, we may realize that $(x^2 - 1)^{\sin x} = e^{\sin(x) \ln(x^2 - 1)}$ and proceed by using the chain rule followed by the product rule.

3. (5 points). Boyle's Law for an ideal gas at constant temperature with pressure P and volume V is $PV = C$, where C is some positive constant. Our gas has $C = 104 \text{ kPa} \cdot \text{m}^3$. Suppose at $t = 10 \text{ s}$, $V = 2 \text{ m}^3$, $P = 52 \text{ kPa}$, and $\frac{dV}{dt} = -\frac{1}{2} \text{ m}^3/\text{s}$. What is $\frac{dP}{dt}$ at this time?

We want to find $\frac{dP}{dt}$, so therefore we must take the derivative of some relation involving P . The only relation we have is $PV = C$, so let us take the derivative with respect to t , recognizing that C is a constant:

$$P \frac{dV}{dt} + \frac{dP}{dt} V = 0.$$

Then we may substitute values we know:

$$52 \cdot \left(-\frac{1}{2}\right) + \frac{dP}{dt} \cdot 2 = 0,$$

and therefore $\frac{dP}{dt} = \frac{52}{2 \cdot 2} = 13 \text{ kPa/s}$.

Extra credit. (2 points). Suppose $y(x)$ is a function satisfying $xy'' + y' + xy = 0$ for all values x , and $y(0) = 1$. Find $y'(0)$ and $y''(0)$. (This y is called a Bessel function of order 0).

Substitute $x = 0$, and then we have $0 \cdot y''(0) + y'(0) + 0 \cdot y(0) = 0$, and so $y'(0) = 0$. Take the implicit derivative with respect to x to get $(xy''' + y'') + y'' + (xy' + y) = 0$, into which we substitute $x = 0$ to get $0 \cdot y'''(0) + 2y''(0) + 0 \cdot y'(0) + y(0) = 0$, which means $2y''(0) + 1 = 0$, and therefore $y''(0) = -\frac{1}{2}$.